

# Copper-Catalyzed Regio- and Enantioselective Hydroboration of Difluoroalkyl-Substituted Internal Alkenes

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Dedicated to Professor Ilan Marek on the occasion of his 60th birthday

Catalytic asymmetric hydroboration of fluoroalkyl-substituted alkenes is a straightforward approach to access chiral small molecules possessing both fluorine and boron atoms. However, enantioselective hydroboration of fluoroalkyl-substituted alkenes without fluorine elimination has been a long-standing challenge in this field. Herein, a copper-catalyzed hydroboration of difluoroalkyl-substituted internal alkenes with high levels of regio- and enantioselectivities is reported. The native carbonyl directing group, copper hydride system, and bisphosphine ligand play crucial roles in suppressing the undesired fluoride elimination. This atom-economic protocol provides a practical synthetic platform to obtain a wide scope of enantioenriched secondary boronates bearing the difluoromethylene moieties under mild conditions. Synthetic applications including functionalization of biorelevant molecules, versatile functional group interconversions, and preparation of difluoroalkylated Terfenadine derivative are also demonstrated.

1. Introduction

Organofluorine compounds have found numerous applications in a broad range of disciplines, thus continuing to drive the

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development of new strategies for their efficient and precise synthesis.[1] In particular, difluoromethylene-featured units are greatly desired structures owing to their prevalence as key scaffolds in a large number of bioactive molecules such as glaucoma treat drug Talfuprost and hepatitis C treatment drug Glecaprevir (Figure 1a).[2] However, compared with the remarkable progresses in the preparation of fluorinated and trifluoromethylated molecules,[3] available synthetic approaches to access difluoromethylene-containing compounds are still relatively limited, especially in a stereoselective manner.[4] Therefore, the development of new catalytic asymmetric transformations to produce diverse difluoromethylene-containing chiral molecules is still highly sought-after.

Catalytic asymmetric hydroboration of unsaturated carboncarbon double bonds represents one of the most renowned reactions for the construction of synthetically useful chiral alkyl boronates.<sup>[5]</sup> In particular, when fluorine-containing alkenes are employed as the substrates, an attractive type of fluorinated organoboron compounds could be obtained.<sup>[6]</sup> The introduction of boryl group not only provides a powerful synthetic handle for versatile downstream transformations but also highlights the significance of fluorine and boron themselves as important moieties in drug design and development.[1,7] Over the past decades, tremendous efforts have been focused on the borylation of trifluoromethyl alkenes.[8,9] In sharp contrast, little progress has been made in the manipulation of difluoroalkylsubstituted alkenes with boron reagents, and only one report by Ito and Hoveyda has addressed the copper-catalyzed enantioselective defluoroborylation of difluoroalkyl-substituted alkenes to give monofluoro-allylic boronates. The proposed mechanism for this selective defluorination stems from the  $\alpha$ -regional region region region that  $\alpha$ -regional region region  $\alpha$ -regional region region  $\alpha$ -regional region region region region region region  $\alpha$ -regional region regio Cu-B complex across the C=C bond to generate difluoroalkylcopper intermediate that is prone to undergo  $\beta$ -fluoride elimination (Figure 1b).[10] However, to the best of our knowledge, catalytic asymmetric hydroboration of difluoroalkylated internal alkenes while keeping the difluoromethylene motif intact is still not realized. The greatest challenge mainly lies in the regioselectivity issue, that is the  $\alpha$ -selective hydrocupration is electronically favored for Cu–H catalysis, [11] but the formed  $\alpha$ -cuprated difluoroalkyl intermediate would easily undergo  $\beta$ -fluoride

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a) Bioactive molecules containing the difluoromethylene moiety

b) Enantioselective defluoroborylation of difluoroalkylated alkenes: Ito & Hoveyda

c) Enantioselective hydroboration of difluoroalkylated internal alkenes: This work

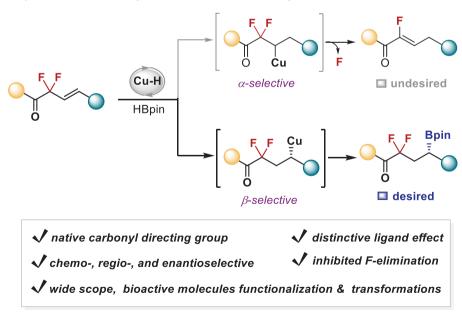


Figure 1. Catalytic asymmetric hydroboration of difluoroalkylated alkenes.



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Table 1. Reaction development.

	EtO Ph + HBpin CuOAc (5 mol% (S,S)-Ph-BPE (5.5 r cyclohexane, rt, 4	mol%) EtO Ph	EtO Ph	EtO Ph
Entry <sup>a)</sup>	Derivations from standard conditions	Ratio of <b>2a/2a</b> '/ <b>2a</b> ''	Yield of <b>2a</b> (%)	ee of <b>2a</b> (%)
1	none	>99:0:0	84(78)	99
2	L2 instead of L1	nd	nr	nd
3	L3 instead of L1	nd	nr	nd
4	L4 instead of L1	47:47:6	trace	nd
5	L5 instead of L1	nd	nr	nd
6 <sup>b)</sup>	L6 instead of L1	10:87:38	trace	nd
7	L7 instead of L1	90:8:2	63	90
8	L8 instead of L1	>99:0:0	68	-89
9	Cu(OAc) <sub>2</sub> instead of CuOAc	72:4:24	18	95
10	CuTc instead of CuOAc	94:0:6	28	99
11	Co(OAc) <sub>2</sub> instead of CuOAc	0:30:70	nd	nd
12	Ni(cod) <sub>2</sub> instead of CuOAc	0:48:52	nd	nd
13	THF instead of cyclohexane	92:6:2	86	59
14	dioxane instead of cyclohexane	82:0:18	70	93
15	toluene instead of cyclohexane	88:2:10	70	93
16	50 °C instead of rt	>99:0:0	85	95

a) Reaction conditions unless otherwise noted: **1a** (0.1 mmol), HBpin (1.5 equiv), CuOAc (5 mol%), ligand (5.5 mol%), solvent (1.0 mL), rt, 48 h. Ratios of **2a**:**2a**':**2a**'' were determined by <sup>19</sup>F-NMR of crude reaction mixtures with PhOCF<sub>3</sub> as the internal standard. Yields of **2a** were determined by <sup>19</sup>F-NMR of crude reaction mixtures with PhOCF<sub>3</sub> as the internal standard, and yield of **2a** in parentheses was determined by the isolated product after column chromatograph. Enantiomeric excess (ee) values of **2a** were determined by chiral HPLC analysis. b) The corresponding carboxylic acid of **2a**' was obtained as the major product. "nd" means "not determined". "nr" means "no reaction".

elimination (Figure 1c, undesired pathway). Therefore, here we report our results on the copper-catalyzed  $\beta$ -regioselective hydroboration of difluoroalkyl-substituted internal alkenes without defluorination to give the chiral difluoroalkyl boronates with high enantioselectivities (Figure 1c, desired pathway). The key findings to suppress the competing fluoride elimination consists of the use of native carbonyl directing group, harnessing Cu–H catalysis system, and the judicious choice of Ph-BPE as the crucial chiral ligand.

# 2. Results and Discussion

### 2.1. Reaction Optimizations

We began our investigation by using the easily accessible (E)–2,2-difluoro-4-phenylbut-3-enoate 1a as the model substrate to undergo hydroboration with HBpin under copper catalysis conditions (Table 1). The screening experiments revealed that the desired hydroboration product 2a was obtained in 78% isolated yield with 99% ee by utilizing CuOAc in combination with Ph-BPE as the chiral ligand, while the defluorinative side-product 2a and the reduction side-products 2a" were not observed (2a/2a'/2a'' > 99:0:0, entry 1). As discussed above, the formation of boronate 2a was attributable to the  $\beta$ -regioselective hydrocupration, while monofluoro alkene 2a was ascribed to the

defluorination process relevant to  $\alpha$ -regioselectivity, thus demonstrating the critical role of  $\beta$ -regionelectivity on suppressing the undesired fluorine elimination. Notably, the bisphosphine ligand plays a critical role in controlling both the reactivity and enantioselectivity (entries 2-8 and Figure 2). For example, the change of Ph-BPE into <sup>i</sup>Pr-BPE, Duphos, Duanphos, or BINAP, almost completely failed to catalyze the reaction (entries 2-5). Interestingly, bulky DTBM-Segphos could promote the reaction of 1a with HBPin, but defluorinative product 2a' was observed to be the major product (entry 6). QuinoxP\* and Josiphos proved to be effective in enabling the selective formation of benzylic boronate 2a, albeit in moderate yields with decreased ee values (entries 7 and 8). The replacement of CuOAc with alternative copper salts such as Cu(OAc), and CuTc resulted in the formation of 2a with 95%-99% ee, but with a significant drop in yields (entries 9 and 10). In comparison, other earth-abundant metal catalysts such as cobalt and nickel, proved to be invalid for this transformation (entries 11 and 12). A screening of reaction media confirmed the optimal choice of cyclohexane as the solvent (entries 13-15). In addition, the yield of 2a was further improved to 85% by operating the reaction at elevated temperatures (50 °C), albeit at the cost of slightly decreased enantioselectivity (95% ee, entry 16). It should be mentioned that subjecting 1a to the optimal Cu-B system reported in Ito & Hoveyda's study led to the formation of difluoro-remaining product B1a in 87% yield with 55% ee (equation 1). $^{[10]}$ 



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Figure 2. Chiral ligands used in this study.

#### 2.2. Substrate Scope

This catalytic enantioselective hydroboration platform proved to be widely compatible to a series of difluoroalkyl-substituted internal alkenes without the loss of fluorine atom (Figure 3). For example, the aryl moieties substituted by various alkyl groups, alkoxyl groups, acetoxyl group, and phenyl group, at different positions of the benzene ring (para, meta, ortho), were all well tolerated under standard conditions to give the corresponding difluoroalkyl boronates **2b–2j** in moderate to high yields with excellent enantioselectivities. Heteroatoms including fluorine, chlorine, sulfur, and silicon, were all compatible in this reaction and the corresponding difluoroalkyl boronates 2k-2o were obtained in good results (94%-99% ee). Incorporation of strong electronwithdrawing carboxylic ester group at the meta position of the benzene ring proved to be viable, affording 2p in good yield with 96% ee. 2-Naphthyl-derived alkenyl difluoroacetate reacted with HBpin to provide 2q with 95% ee. Furthermore, heteroaromatic groups such as carbazole and pyridine were also accommodated under identical conditions to give hydroboration products 2r and 2s in good yields with 99% and 89% ee, respectively. In addition, the scope of this reaction also exhibited some limitations, for example, substrate bearing strong electron-withdrawing group at the para position of phenyl ring, trisubstituted alkene, and alkyl group-substituted difluoroalkyl alkene were inert under current

With respect to the difluoroalkyl substituent (**Figure 4**), switching ethyl group to more sterically hindered *tert*-butyl group was found to be well acceptable (**4a**). Furthermore, this reaction protocol is not limited to alkenyl difluoroacetates, and a broad range of tertiary difluoroacetamides are also good reaction partners. For example, dimethyl and dibenzyl-containing amides underwent  $\beta$ -regioselective hydroboration smoothly (**4b** and **4c**). The abso-

lute configuration of boronate **4b** was determined to be *S* by X-ray crystallographic analysis, and other products were assigned by analogy. A series of cyclic amides, including pyrrole, piperidine, morpholine, piperazine, and azepane-containing substrates, reacted with HBpin under the standard conditions to give difluoroalkyl boronates **4d–4h** in good yields with high enantioselectivities (90%–94% ee). More importantly, alkenyl difluoroalkyl acetoxyl substrates also underwent the desired hydroboration, giving the corresponding boronates **4i–4k** in high yields and high enantioselectivities.

The functional group compatibility and potential utility of this method was further demonstrated by the functionalization of biorelevant molecules (Figure 5). Phenylalanine-derived alkenyl difluoroacetate derivative underwent chemo-, regio- and stereos-elective hydroboration without fluorine elimination to provide 6a in a practical yield with high dr value. Natural product structures including Borneol and Estradiol were amenable to be equipped at the carbonyl and aryl sites, affording boronates 6b and 6c with excellent diastereoselectivities. Furthermore, three pharmaceutical molecules including attention deficit hyperactivity disorder drug Atomoxetine, nonsteroidal anti-inflammatory drug Ibuprofen, and antihyperlipidemic drug Fenofibrate were also successfully decorated with the difluoroalkyl boronate motif (6d–6f).

# 2.3. Synthetic Transformations

Subsequently, three gram-scale reactions were conducted to probe the practicality of this transformation to furnish the desired products 2a, 4b, and 4i in good yields with maintained high level of enantioselectivies (Figure 6a). More importantly, the obtained chiral difluoroalkyl boronates could undergo a series of boron-based stereospecific functional group interconversions (Figure 6b). Among them, oxidation of 4i by NaBO<sub>3</sub> led to *gem*-difluorinated 1,4-diol 7 in 75% yield with full stereoretention.<sup>[13]</sup> Treating 4i with 2-thienyllithium in the presence of *N*-bromosuccinimide provided a thienylated difluoroalkyl ester 8 in 73% yield without loss of ee values.<sup>[14]</sup> Iodine-promoted vinylation of boronate ester proceeded with conservation of

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2j, 57%, 94% ee



2m, 61%, 97% ee





**2o**, 73%, 98% ee

2p, 75%, 96% ee



2q, 51%, 95% ee

2r, 75%, 99% ee

2s, 67%, 89% ee

no reaction

low conversion

no reaction

Figure 3. Substrate scope of alkenyl difluoroacetates.

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**Figure 4.** Substrate scope with respect to the difluoroalkyl substituent. <sup>a</sup> The corresponding alcohols were obtained via oxidation of the crude boronates by NaBO<sub>3</sub>.

enantiomeric excess to give the corresponding difluorinated alcohol 9 in 79% yield.[15] Palladium-catalyzed Suzuki-Miyaura cross-coupling reaction of boronate 4i with iodioanisole was also viable to generate compound 10 in high yield with predominant stereo-retention at the chiral center.[16] Furthermore, Matteson homologation of secondary boronic ester 4i occurred without erosion of optical purity, thus yielding compound 11 in 73% yield with 98% ee.[17] More importantly, this protocol was further applied in the asymmetric synthesis of difluoroalkyl analogue of antihistamine drug Terfenadine (Figure 6c).[18] Copper-catalyzed enantioselective hydroboration of 12 yielded the boronate as an intermediate, which was subsequently oxidized to chiral alcohol 13 in 81% total yield with 98% ee. Further reduction of 13 with borane provided the corresponding amine 14 in 47% total yield through a three-step procedure. In addition, it should be mentioned that an alternative path to access the compounds in Figure 6b by performing a cascade sequence of enantioselective conjugate addition to enones and deoxyfluorination transformations might also be possible.[19]

#### 2.4. Mechanistic Studies

A series of experiments were performed to gain insight on the mechanism of this reaction (Figure 7). First, isotope labelling

experiment with D-Bpin under otherwise standard conditions generated the corresponding D-2a with 95% deuterium incorporation in 61% yield and 99% ee (Figure 7a). The excellent diastereoselectivity of D-2a suggested an irreversible addition of Cu-H species across the C=C bond and the hydrocupration process is thus estimated to be the regio- and stereo-determining step of this reaction. Subsequently, we conducted a series of control experiments with other types of substrates to understand the influence of alkene structures. As illustrated in Figure 7b, both the Z-configured isomer of model substrate 1a and the 1,1-disubstituted alkenyl difluoroacetate 15, was nearly inert under standard conditions ( $\leq$  5% conversions). Interestingly, substrates without the difluoromethylene moiety (16), underwent this copper-catalyzed hydroboration with constant exclusive  $\beta$ regioselectivity, albeit in decreased yield and enantioselectivity to some extent (product 17). In sharp contrast, both of 1,2disubstituted trifluoromethyl styrene 18 and difluoroalkyl ether 19 were found to be almost unreactive under otherwise identical conditions. Collectively, these control experiments clearly demonstrated the crucial role of carbonyl scaffold as the native directing group in improving the reactivity and dictating regioselectivity. Based on these results and previous studies, [20] a plausible mechanism is illustrated in Figure 7c. The reaction starts with the formation of ligated copper hydride species (\*LCu-H) from copper salt, bisphosphine ligand, and HBpin.  $\beta$ -Regioselective

Figure 5. Applications for the functionalization of biorelevant molecules.

hydrocupration undergoes stereoselectively to give the chiral alkylcopper complex Int1. Final borylation occurs with retention of configuration via  $\sigma$ -bond metathesis to produce the desired product and regenerates the \*Cu–H catalyst. Notably, the use of copper hydride catalysis system, the choice of Ph-BPE as the chiral ligand, and the presence of native carbonyl directing group all plays indispensable roles in enabling the realization of this atom-economic reaction.

To dig deeper into the mechanism and origin of the regio- and stereo-selectivities, density functional theory calculations have been carried out on the reaction between 1a and HBpin catalyzed by L1-Cu complex (see Supporting Information for more details). As shown in Figure 8A, copper hydride complex IM1, as the zero point, inserts into alkene 1a to generate IM2 via TS1 exogenously by 8.8 kcal  $\mathrm{mol}^{-1}$  with an activation free energy of 11.3 kcal  $\mathrm{mol}^{-1}$ . And then,  $\sigma$ -bond metathesis between H–B bond of HBpin and Cu–C bond of IM2 occurs to release the final product 2a and regenerate the active Cu-hydride catalyst IM1, which requires an energy barrier of 19.5 kcal  $\mathrm{mol}^{-1}$  via TS2. The calculation results show that the  $\sigma$ -bond metathesis (TS2) is identified as the turnover-limiting step, and moreover, the irreversible alkene insertion step (TS1) exercises control over the regio- and stereo-selectivities, which is consistent with our deuterium studies.

With comprehension of the reaction mechanism, we next sought to scrutinize the observed selectivities. As shown in Figure 8B, the activation barrier for 2,1-migratory insertion of 1a via **TS1** (β-regioselectivity) is 4.0 kcal mol<sup>-1</sup> more favorable than that for 1,2-migratory insertion into via **TS1**" ( $\alpha$ -regioselectivity), which is in accord with the experimentally observed regioselectivity. In addition, the enantioselectivity is controlled by the relative energy (9.5 kcal mol<sup>-1</sup>) of the two diastereomeric transition states TS1 (leading to S-2a) and TS1' (leading to R-2a) for the insertion of 1a. Noncovalent interactions (NCIs) using the IGMH analysis has been adopted to comparing the three competing transition states. Obviously, TS1 has more pronounced C-H···H-C (contributed by the carbonyl group) and  $C-H\cdots\pi$  dispersion interactions between 1a and Cu catalyst compared to TS1' and TS1", which accounts for the observed selectivities of the Cu-catalyzed hydroboration reaction.

#### 3. Conclusion

In conclusion, we have successfully developed a regio- and enantioselective hydroboration of difluoroalkyl-substituted internal alkenes without the loss of fluorine atom. The key to the success of this reaction relies on the concerted engagement of

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#### a) Gram-scale reactions

## b) Synthetic transformations

Reaction conditions: A) NaBO<sub>3</sub>•4H<sub>2</sub>O (1.5 equiv.), THF/H<sub>2</sub>O(1:1), rt, 4 h; then K<sub>2</sub>CO<sub>3</sub> (2 equiv.), MeOH, rt, 12 h; B) 2-thiophenyllithium (1.1 equiv.), NBS (2.2 equiv), -78 °C to rt, 3 h; C) vinylmagnesium bromide (1.1 equiv.), I<sub>2</sub>, NaOMe, -78 °C to rt, 3 h; D) Pd(PPh<sub>3</sub>)<sub>4</sub> (8 mol%), PPh<sub>3</sub> (32 mol%), p-iodoanisole (1.2 equiv.), K<sub>2</sub>CO<sub>3</sub> (1 equiv.), Ag<sub>2</sub>O (1 equiv.), DME, 80 °C, 24 h; E) CICH<sub>2</sub>Br (2 equiv),  $^n$ BuLi (1.2 equiv), THF, -78 °C to rt, 3 h

# c) Synthesis of Terfenadine derivative

Figure 6. Further demonstration of synthetic utility.

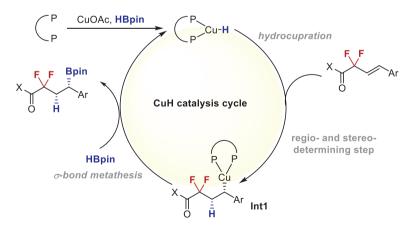
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# a) Deuterium experiment

#### b) Control experiments with other type of substrates

# c) Proposed mechanism



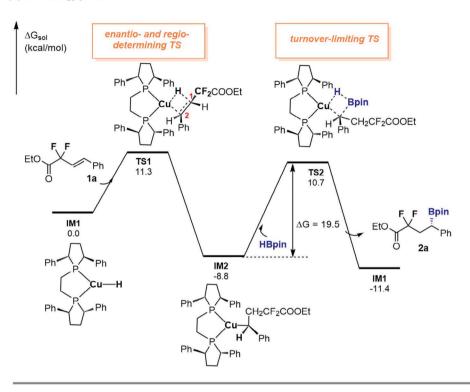
☐ Key for the regioselectivity: Native carbonyl directing group and Ph-BPE as the ligand

Figure 7. Experimental mechanistic studies.

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# (A) Free energy profile



#### (B) Non-covalent interactions analysis for the three migratory insertion transition states

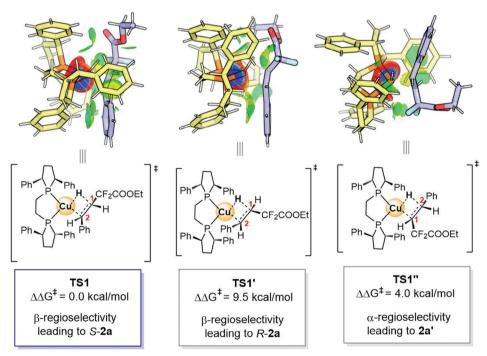


Figure 8. Computational mechanistic studies.



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native carbonyl directing group, copper hydride catalysis system, and Ph-BPE as the chiral ligand. This atom-economic protocol provides practical access to a wide range of chiral difluoroalkyl boronates that are viable to versatile further synthetic transformations with high stereochemical fidelity. Combined experimental and computational studies indicate that hydrocupration process is the regio- and stereo-selectivity determining step, and the carbonyl-containing difluoroalkyl substrates manifest important noncovalent interactions to stabilize the corresponding transition states. We anticipate this unprecedented strategy will find broad applications in both synthetic chemistry and drug development. Further investigation into the mechanism and application to other substrates are currently underway in our laboratories.

[CCDC 2 262 049 (4b) contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data\_request/cif.]

# **Supporting Information**

Supporting Information is available from the Wiley Online Library or from the author.

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# **Conflict of Interest**

The authors declare no conflict of interest.

# **Data Availability Statement**

The data that support the findings of this study are available in the supplementary material of this article.

#### **Keywords**

copper catalysis, difluoromethylene group, enantioselectivity, hydroboration, regioselectivity

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a) W. K. Hagmann, J. Med. Chem. 2008, 51, 4359; b) S. Purser,
 P. R. Moore, S. Swallow, V. Gouverneur, Chem. Soc. Rev. 2008, 37,
 320; c) J. Wang, M. Sánchez-Roselló, J. L. Aceña, C. Del Pozo, A. E.
 Sorochinsky, S. Fustero, V. A. Soloshonok, H. Liu, Chem. Rev. 2014,
 114, 2432; d) E. P. Gillis, K. J. Eastman, M. D. Hill, D. J. Donnelly, N.
 A. Meanwell, J. Med. Chem. 2015, 58, 8315; e) M. Inoue, Y. Sumii, N.

- Shibata, ACS Omega 2020, 5, 10633; f) R. Britton, V. Gouverneur, J.-H. Lin, M. Meanwell, C. Ni, G. Pupo, J.i-C. Xiao, J. Hu, Nat. Rev. Meth. Primers 2021, 1, 47; g) C. J. Hawker, A. K. Whittaker, Chem. Rev. 2022, 122, 167.
- [2] a) C. S. Burgey, K. A. Robinson, T. A. Lyle, P. E. J. Sanderson, S. D. Lewis, B. J. Lucas, J. A. Krueger, R. Singh, C. Miller-Stein, R. B. White, B. Wong, E. A. Lyle, P. D. Williams, C. A. Coburn, B. D. Dorsey, J. C. Barrow, M. T. Stranieri, M. A. Holahan, G. R. Sitko, J. J. Cook, D. R. Mcmasters, C. M. Mcdonough, W. M. Sanders, A. A. Wallace, F. C. Clayton, D. Bohn, Y. M. Leonard, T. J. Detwiler, J. J. Lynch, Y. Yan, et al., J. Med. Chem. 2003, 46, 461; b) T. Nakajima, T. Matsugi, W. Goto, M. Kageyama, N. Mori, Y. Matsumura, H. Hara, Biol. Pharm. Bull. 2003, 26, 1691; c) W. J. Hoekstra, E. P. Garvey, W. R. Moore, S. W. Rafferty, C. M. Yates, R. J. Schotzinger, Bioorg. Med. Chem. Lett. 2014, 24, 3455; d) S. Zeuzem, G. R. Foster, S. Wang, A. Asatryan, E. Gane, J. J. Feld, T. Asselah, M. Bourlière, P. J. Ruane, H. Wedemeyer, S. Pol, R. Flisiak, F. Poordad, W.-L. Chuang, C. A. Stedman, S. Flamm, P. Kwo, G. J. Dore, G. Sepulveda-Arzola, S. K. Roberts, R. Soto-Malave, K. Kaita, M. Puoti, J. Vierling, E. Tam, H. E. Vargas, R. Bruck, F. Fuster, S.-W. Paik, F. Felizarta, et al., N. Engl. J. Med. 2018, 378,
- [3] a) J.-A. Ma, D. Cahard, Chem. Rev. 2008, 108, PR1; b) D. Cahard, X. Xu, S. Couve-Bonnaire, X. Pannecoucke, Chem. Soc. Rev. 2010, 39, 558; c) J. Nie, H.-C. Guo, D. Cahard, J.-A. Ma, Chem. Rev. 2011, 111, 455; d) X. Yang, T. Wu, R. J. Phipps, F. D. Toste, Chem. Rev. 2015, 115, 826; e) Y.i Zhu, J. Han, J. Wang, N. Shibata, M. Sodeoka, V. A. Soloshonok, J. A. S. Coelho, F. D. Toste, Chem. Rev. 2018, 118, 3887; f) F.-L. Qing, X.-Y. Liu, J.-A. Ma, Q. Shen, Q. Song, P. Tang, CCS Chem 2022, 4, 2518.
- [4] a) J.-S. Yu, F.-M. Liao, W.-M. Gao, K. Liao, R.-L. Zuo, J. Zhou, Angew. Chem., Int. Ed. 2015, 54, 7381; b) S. M. Banik, J. W. Medley, E. N. Jacobsen, Science 2016, 353, 51; c) X.-S. Hu, J.-S. Yu, J. Zhou, Chem. Commun. 2019, 55, 13638; d) X. Gao, R. Cheng, Y.-L. Xiao, X.-L. Wan, X. Zhang, Chem 2019, 5, 2987; e) M.-Y. Rong, J.-S. Li, Y. Zhou, F.-G. Zhang, J.-A. Ma, Org. Lett. 2020, 22, 9010; f) E. Miller, S. Kim, K. Gibson, J. S. Derrick, F. D. Toste, J. Am. Chem. Soc. 2020, 142, 8946; g) S. Huang, F.-F. Tong, D.-C. Bai, G.-P. Zhang, Y.-J. Jiang, B. Zhang, X. Leng, Y.-L. Guo, X.-L. Wan, X. Zhang, C.-H. Ding, X.-L. Hou, Nat. Commun. 2021, 12, 6551; h) H. Uno, K. Kawai, T. Araki, M. Shiro, N. Shibata, Angew. Chem. 2022, 134, e202117635.
- [5] a) K. Burgess, M. J. Ohlmeyer, Chem. Rev. 1991, 91, 1179; b) C. M. Crudden, D. Edwards, Eur. J. Org. Chem. 2003, 4695; c) J. V. Obligacion, P. J. Chirik, Nat. Rev. Chem. 2018, 2, 15; d) X. Yang, S. Ge, S. Curr. Opin. Green Sustainable Chem. 2021, 31, 100542; e) S. J. Geier, C. M. Vogels, J. A. Melanson, S. A. Westcott, Chem. Soc. Rev. 2022, 51, 8877; f) H. Lu, B. Li, Chin. J. Org. Chem. 2022, 42, 3167.
- [6] X. Ma, Z. Kuang, Q. Song, JACS Au 2022, 2, 261.
- [7] a) R. Smoum, A. Rubinstein, V. M. Dembitsky, M. Srebnik, Chem. Rev. 2012, 112, 4156; b) D. B. Diaz, A. K. Yudin, Nat. Chem. 2017, 9, 731;
  c) C. Sandford, V. K. Aggarwal, Chem. Commun. 2017, 53, 5481; d) H. Wang, C. Jing, A. Noble, V. K. Aggarwal, Angew. Chem., Int. Ed. 2020, 59, 16859.
- [8] a) P. V. Ramachandran, M. P. Jennings, H. C. Brown, Org. Lett. 1999, 1, 1399; b) P. V. Ramachandran, M. P. Jennings, Org. Lett. 2001, 3, 3789; c) T. Braun, M. Ahijado Salomon, K. Altenhöner, M. Teltewskoi, S. Hinze, Angew. Chem., Int. Ed. 2009, 48, 1818; d) Z. Kuang, H. Chen, J. Qiu, Z. Ou, Y. Lan, Q. Song, Chem 2020, 6, 2347; e) J. Li, C. Liu, J. He, S. Xu, X. Zhao, Y. Zhu, S. Cao, Org. Chem. Front. 2020, 7, 1495; f) Y. Kojima, Y. Nishii, K. Hirano, Org. Lett. 2022, 24, 7450. g) Q. Jiang, T. Guo, Z. Yu, J. Org. Chem. 2017, 82, 1951; h) B. Liu, H.-H. Wu, J. Zhang, ACS Catal. 2018, 8, 8318; i) C. Yang, Z.-L. Liu, D.-T. Dai, Q. Li, W.-W. Ma, M. Zhao, Y.-H. Xu, Org. Lett. 2020, 22, 1360; j) J. Wu, H. Wu, X. Li, X. Liu, Q. Zhao, G. Huang, C. Zhang, Angew. Chem., Int. Ed. 2021, 60, 20376; k) M. Hu, B. B. Tan, S. Ge, J. Am. Chem. Soc. 2022, 144, 15333.



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- [9] a) Y. Huang, T. Hayashi, J. Am. Chem. Soc. 2016, 138, 12340; b) M. Wang, X. Pu, Y. Zhao, P. Wang, Z. Li, C. Zhu, Z. Shi, J. Am. Chem. Soc. 2018, 140, 9061; c) P. Gao, C. Yuan, Y. Zhao, Z. Shi, Chem 2018, 4, 2201; d) R. Kojima, S. Akiyama, H. Ito, Angew. Chem., Int. Ed. 2018, 57, 7196; e) P. H. S. Paioti, J. Del Pozo, M. S. Mikus, J. Lee, M. J. Koh, F. Romiti, S. Torker, A. H. Hoveyda, J. Am. Chem. Soc. 2019, 141, 19917; f) T. W. Butcher, J. L. Yang, J. F. Hartwig, Org. Lett. 2020, 22, 6805; g) Z. Fan, M. Ye, Y. Wang, J. Qiu, W. Li, X. Ma, K. Yang, Q. Song, ACS Cent. Sci. 2022, 8, 1134.
- [10] S. Akiyama, K. Kubota, M. S. Mikus, P. H. S. Paioti, F. Romiti, Q. Liu, Y. Zhou, A. H. Hoveyda, H. Ito, Angew. Chem., Int. Ed. 2019, 58, 11998.
- a) C. Deutsch, N. Krause, B. H. Lipshutz, Chem. Rev. 2008, 108, 2916;
   b) B. Lipshutz, Synlett 2009, 4, 509;
   c) R. Y. Liu, S. L. Buchwald, Acc. Chem. Res. 2020, 53, 1229.
- [12] a) S. M. Smith, J. M. Takacs, J. Am. Chem. Soc. 2010, 132, 1740; b) Y. Xi, J. F. Hartwig, J. Am. Chem. Soc. 2016, 138, 6703; c) S. Chakrabarty, H. Palencia, M. D. Morton, R. O. Carr, J. M. Takacs, Chem. Sci. 2019, 10, 4854; d) T.-T. Gao, W.-W. Zhang, X. Sun, H.-X. Lu, B.-J. Li, J. Am. Chem. Soc. 2019, 141, 4670; e) W. Zhao, K.-Z. Chen, A.-Z. Li, B.-J. Li, J. Am. Chem. Soc. 2022, 144, 13071.
- [13] G. W. Kabalka, T. M. Shoup, N. M. Goudgaon, J. Org. Chem. 1989, 54, 5930.

- [14] A. Bonet, M. Odachowski, D. Leonori, S. Essafi, V. K. Aggarwal, Nat. Chem. 2014, 6, 584.
- [15] a) G. Zweifel, H. Arzoumanian, C. C. Whitney, J. Am. Chem. Soc. 1967, 89, 3652; b) D. A. Evans, T. C. Crawford, R. C. Thomas, J. A. Walker, J. Org. Chem. 1976, 41, 3947.
- [16] D. Imao, B. W. Glasspoole, V. S. Laberge, C. M. Crudden, J. Am. Chem. Soc. 2009, 131, 5024.
- [17] K. M. Sadhu, D. S. Matteson, Organometallics 1985, 4, 1687.
- [18] J. I. Perlmutter, L. T. Forbes, D. J. Krysan, K. Ebsworth-Mojica, J. M. Colquhoun, J. L. Wang, P. M. Dunman, D. P. Flaherty, J. Med. Chem. 2014. 57, 8540.
- [19] a) F. Beaulieu, L.-P. Beauregard, G. Courchesne, M. Couturier, F. Laflamme, A. L'heureux, Org. Lett. 2009, 11, 5050; b) H. Wu, S. Radomkit, J. M. O'Brien, A. H. Hoveyda, J. Am. Chem. Soc. 2012, 134, 8277; c) S. Kobayashi, P. Xu, T. Endo, M. Ueno, T. Kitanosono, Angew. Chem., Int. Ed. 2012, 51, 12763; d) R. Kucera, F. W. Goetzke, S. P. Fletcher, Org. Lett. 2020, 22, 2991.
- [20] a) D. Noh, H. Chea, J. Ju, J. Yun, Angew. Chem., Int. Ed. 2009, 48, 6062;
  b) D. Noh, S. K. Yoon, J. Won, J. Y. Lee, J. Yun, Chem. Asian J. 2011, 6, 1967;
  c) Y. Xi, J. F. Hartwig, J. Am. Chem. Soc. 2017, 139, 12758;
  d) X. Yang, C. Yuan, S. Ge, Chem 2022, 9, 198.